

9.2. For this problem, we integrate (9.2.3) by parts and write

$$a_2 \lambda^3 = -\frac{2\pi}{3kT} \int_0^\infty e^{-u(r)/kT} \frac{\partial u(r)}{\partial r} r^3 dr ;$$

cf. eqn. (3.7.17) and Problem 3.23. With given $u(r)$, we get

$$\begin{aligned} a_2 \lambda^3 &= \frac{2\pi}{3kT} \int_0^\infty e^{-A/kTr^m} e^{B/kTr^n} \left(\frac{mA}{r^{m-2}} - \frac{nB}{r^{n-2}} \right) dr \\ &= \frac{2\pi}{3kT} \int_0^\infty e^{-A/kTr^m} \sum_{j=0}^\infty \frac{1}{j!} \left(\frac{B}{kT} \right)^j \left(\frac{mA}{r^{m-2+nj}} - \frac{nB}{r^{n-2+nj}} \right) dr \\ &= \frac{2\pi}{3kT} \sum_{j=0}^\infty \frac{1}{j!} \left(\frac{B}{kT} \right)^j \left\{ A \Gamma \left(\frac{m-3+nj}{m} \right) \left(\frac{kT}{A} \right)^{(m-3+nj)/m} \right. \\ &\quad \left. - \frac{n}{m} B \Gamma \left(\frac{n-3+nj}{m} \right) \left(\frac{kT}{A} \right)^{(n-3+nj)/m} \right\}. \end{aligned}$$

From the first sum we take the ($j=0$)-term out and combine the remaining terms with the second sum (in which the index j is changed to $j-1$); after considerable simplification, we get

$$a_2 \lambda^3 = \frac{2\pi}{3} \left(\frac{A}{kT} \right)^{3/m} \left\{ \Gamma \left(\frac{m-3}{m} \right) - \frac{3}{m} \sum_{j=1}^\infty \frac{1}{j!} \Gamma \left(\frac{nj-3}{m} \right) \left[\frac{B}{kT} \left(\frac{kT}{A} \right)^{n/m} \right]^j \right\}. \quad (1)$$

For comparison with other cases, we set $A = A' r_0^m$ and $B = B' r_0^n$ (so that A' and B' become direct measures of the energy of interaction). Expression (1) then becomes

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$$a_2 \lambda^3 = \frac{2\pi}{3} r_0^3 \left(\frac{A'}{kT} \right)^{3/m} \left\{ \Gamma \left(\frac{m-3}{m} \right) - \frac{3}{m} \sum_{j=1}^\infty \frac{1}{j!} \Gamma \left(\frac{nj-3}{m} \right) \left[\frac{B'}{kT} \left(\frac{kT}{A'} \right)^{n/m} \right]^j \right\}. \quad (2)$$

Now, to simulate a *hard-core* repulsive interaction, we let $m \rightarrow \infty$, with the result that

$$a_2 \lambda^3 = \frac{2\pi}{3} r_0^3 \left\{ 1 - 3 \sum_{j=1}^\infty \frac{1}{(nj-3)j!} \left(\frac{B'}{kT} \right)^j \right\}. \quad (2a)$$

With $n=6$, expression (2a) reduces to the one derived in the preceding problem. Furthermore, if terms with $j > 1$ are neglected, we recover the van der Waals approximation (9.3.8).

For further comparison, we look at the behavior of the coefficient $B_2 (\equiv a_2 \lambda^3)$ at high temperatures. While the hard-core expression (2a) predicts a constant B_2 as $T \rightarrow \infty$, the soft-core expression (2) predicts a B_2 that ultimately vanishes, as $T^{-3/m}$, which agrees qualitatively with the data

$$a_2 \lambda^3 = \frac{2\pi}{3} r_0^3 \left(\frac{A'}{kT} \right)^{3/m} \left\{ \Gamma \left(\frac{m-3}{m} \right) - \frac{3}{m} \sum_{j=1}^{\infty} \frac{1}{j!} \Gamma \left(\frac{nj-3}{m} \right) \left[\frac{B'}{kT} \left(\frac{kT}{A'} \right)^{n/m} \right]^j \right\}. \quad (2)$$

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9.7. To the desired approximation,

$$\frac{P}{kT} \equiv \frac{1}{V} \ln \mathfrak{z} = \frac{1}{\lambda^3} (z - a_2 z^2), \quad n = \frac{N}{V} = \frac{1}{\lambda^3} (z - 2a_2 z^2), \quad (1a,b)$$

where a_2 is the second virial coefficient of the gas. It follows that

$$z = n\lambda^3(1 + 2a_2 \cdot n\lambda^3), \quad \text{whence } P = nkT(1 + a_2 \cdot n\lambda^3). \quad (2a,b)$$

Next

$$A = NkT \ln z - PV = NkT \{ \ln(n\lambda^3) - 1 + a_2 \cdot n\lambda^3 \},$$

$$G = NkT \ln z = NkT \{ \ln(n\lambda^3) + 2a_2 \cdot n\lambda^3 \},$$

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$$S = - \left(\frac{\partial A}{\partial T} \right)_{N,V} = Nk \left\{ \frac{5}{2} - \ln(n\lambda^3) - n \frac{\partial}{\partial T} (Ta_2 \lambda^3) \right\};$$

remember that the coefficient a_2 is a function of T . Furthermore,

$$U = A + TS = NkT \left\{ \frac{3}{2} - nT \frac{\partial}{\partial T} (a_2 \lambda^3) \right\},$$

$$H = U + PV = NkT \left\{ \frac{5}{2} - nT^2 \frac{\partial}{\partial T} \left(\frac{a_2 \lambda^3}{T} \right) \right\},$$

$$C_V = \left(\frac{\partial U}{\partial T} \right)_{N,V} = Nk \left\{ \frac{3}{2} - n \frac{\partial}{\partial T} \left(T^2 \frac{\partial}{\partial T} (a_2 \lambda^3) \right) \right\}, \quad \text{and}$$

$$C_P - C_V = -T \frac{(\partial P / \partial T)_{N,V}^2}{(\partial P / \partial V)_{N,T}} = Nk \left\{ 1 + 2nT \frac{\partial}{\partial T} (a_2 \lambda^3) \right\}.$$

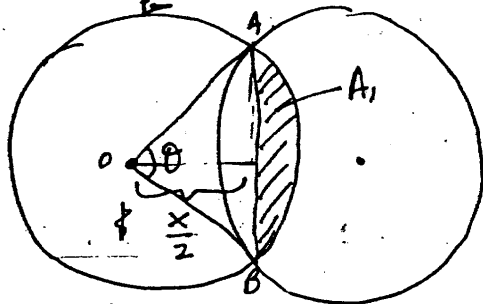
For the second part, use the expression for $a_2 \lambda^3$ derived in Problem 9.5 and examine the temperature dependence of the various thermodynamic quantities.

3. Hard sphere : $u = \begin{cases} a & : r < a/2 \\ 0 & : r > a/2 \end{cases}$ (Let $a/2 = r_0$)

For 3 dimensions problem is solved in pathria ; $a_1 = 1$, $a_2 = \frac{2\pi r_0^3}{3\lambda^3}$, $a_3 = \frac{5}{18} \pi^2 \left(\frac{r}{\lambda}\right)^6$

For 2 DIMENSIONS :

$$= \frac{1}{6\lambda^4} \int_0^{r_0} d^2 r_{12} \left[\text{Area of overlap of the two circles} \right]$$



Area of triangle OAB
 $= 2 \left(\frac{1}{2} r_0^2 \sin \frac{\theta}{2} \cos \frac{\theta}{2} \right)$
 $= \frac{1}{2} r_0^2 \sin \theta$

lets ~~say~~ $r_{12} \equiv x$

Shaded area = $A_1 = \left(\frac{1}{2} r_0^2 \theta - \text{area of triangle} \right)$
 $= \frac{1}{2} r_0^2 (\theta - \sin \theta)$

$$\therefore a_3 = \frac{1}{6\lambda^4} \int_0^{2r_0} 2\pi x dx r_0^2 (\theta - \sin \theta)$$

where ; $\cos \frac{\theta}{2} = x/2r_0$

$$a_3 = \frac{1}{6\lambda^4} \int_0^{\pi} 2\pi r_0^4 d\theta \sin \theta (\theta - \sin \theta)$$

$$= \frac{\pi^2}{6} \left(\frac{r_0}{\lambda}\right)^4$$

a_2 is trivially calculated to be $\frac{\pi r_0^2}{2\lambda^2}$

1 DIM $a_3 = \frac{1}{\lambda} \frac{r_0^2}{2}$, $a_2 = \frac{A_0}{\lambda}$