Welcome to PY106

-The syllabus is your guide to this course. It contains information about the discussions, labs., the class, lab. and

- exam. schedules and grading scheme, etc.
- Discussion sessions begin today!
- Labs. begin on Jan. 28.
- Assignment 1 is a hand-in, and will be posted on Blackboard soon. It is due on Tuesday (Jan. 28) 10:00pm.
- Most other assignments are posted on WebAssign. To
- access the assignment, you need to acquire an access code from WebAssign. Instructions for WebAssign can be found in the syllabus.
- Lecture notes can be downloaded from

http://physics.bu.edu/~okctsui/PY106.html. Note that the URL is case sensitive.

Setting the Channel Number for Your Clicker

- 1. Press and release the "CH" button.
- 2. While the light is flashing red and green, enter the two digit channel code "41" for this class.
- 3. After the second digit is entered, press and release the "CH" button. The light should flash green to confirm.
- 4. Press and release the "1/A" button. The light should flash amber ONCE to confirm. If it flashes continuously, there is probably an error and you should try it again.

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Electric charges

There are two kinds of electric charge, positive and negative.

Objects are generally charged by either acquiring extra electrons (a net negative charge), or giving up electrons (a net positive charge).

Forces between charged objects can be very large. Such forces are really what stop us from falling through the floor. In other words, what we called the normal force is really associated with repulsive forces between electrons.

Our symbol for charge is Q or q. The unit is the coulomb (C).

Observations

Place one nylon rod on a rotating stand, and bring another nylon rod close to the first rod, without allowing the rods to touch.

Nothing happens.

Repeat after rubbing both rods with a piece of silk.

Now the rods repel one another.

How can we explain this? Rubbing a nylon rod with silk transfers electrons from one to the other, giving the rod a net charge. Both rods have a net charge of the same sign, and tike charges repel. $_5$

Observations

When we go through a similar process with rubber rods, using felt or fur, we also observe that the rods repel one another, after they are rubbed with the felt or fur.

What happens when a rubber rod, after being rubbed with felt or fur, is brought close to a nylon rod that has been rubbed with silk?

The rods attract each other.

In the last experiment, we learned that like charges repel. The behavior found here, being opposite to that found in the last experiment must tell us that the signs of the net charges on the rods are opposite. Therefore, unlike charges attract.

Charging an object

An object can acquire a net charge by touching another material. Which material acquires electrons is determined by where the materials fit in the *triboelectric series*.

MORE POSITIVE rabbit's fur glass nylon cat's fur silk polyethylene rubber balloon MORE NEGATIVE	When materials are rubbed together, the one higher up the list gives electrons to the one further down the list.
	Nylon is higher up than silk, so nylon gives up electrons, becoming positive, when rubbed with silk.
	Rubber is lower than fur (or felt), so rubber acquires electrons and a negative charge. 7









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Charge and mass

Charging an object almost always involves transferring electrons. In this case, the object must have lost 6.25×10^{18} electrons, making it lighter.

The charge-to-mass ratio of the electron is:

$$\frac{e}{m} = \frac{1.60 \times 10^{-19} \text{ C}}{9.11 \times 10^{-31} \text{ kg}} = 1.76 \times 10^{11} \text{ C/kg}$$

So the mass loss per 1 C change in charge is the inverse of this, which is:

 $\frac{m}{2} = 5.69 \times 10^{-12} \text{ kg/C}$

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Two spheres

As long as we allow the spheres to touch in a way that no charge is transferred in or out of the system, we can use the idea that charge is conserved – the net charge in the system must be constant at all times.

In this case, the net charge is +7Q - 3Q = +4Q. Because the spheres are metal, electrons can move around easily on the spheres. The spheres are identical, so the net charge of +4Q divides evenly between the spheres, giving each sphere a charge of +2Q.

Classifying materials

We can classify materials into three broad categories, based on how easily charge flows through them.

Conductors: charge flows easily (e.g., metals).

Semi-conductors: charge flows, but not easily (e.g. graphite).

Insulators: very little charge flow (e.g. plastic).

Application: a power cord you plug into a wall socket has two conducting wires to carry electricity to your cell phone and back, but the wires are wrapped with a rubber coating so you do not get a shock.

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A soda can

In both cases, the can is attracted.

Whether the rod is positive or negative, the rod polarizes the can (the can acts like a huge polar molecule) so the side of the can closest to the rod has a charge opposite to the charge on the rod. This leads to a net attractive force between the can and the rod.

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Try this at home

A thin stream of water is another excellent charge detector. Amaze your friends by charging an object by rubbing it, and then attracting the stream of water toward the charged object.

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